Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

# Kathryn Hassall\* and Jonathan White

School of Chemistry, University of Melbourne, Parkville, VIC 3010, Australia

Correspondence e-mail: whitejm@unimelb.edu.au

### Key indicators

Single-crystal X-ray study T = 130 KMean  $\sigma$ (C–C) = 0.003 Å Disorder in solvent or counterion R factor = 0.046 wR factor = 0.126 Data-to-parameter ratio = 11.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# Dimethyl(2-trimethylsilylethyl)[(2-trimethylsilylethyl)dimethylammonio]ammonium tetrakis[3,5-bis(trifluoromethyl)phenyl]borate at 130 K

The crystal structure of the title compound,  $C_{14}H_{39}N_2Si_2^+$ - $C_{32}H_{12}F_{24}B^-$ , reveals a centrosymmetric cation consisting of two 2-(trimethylsilylethyl)dimethylamine molecules held together by a symmetrical  $N \cdots H \cdots N$  hydrogen bond. The N-H distance is 1.347 (2) Å, the  $N \cdots N$  distance is 2.695 (3) Å and the  $N \cdots H \cdots N$  angle is 180°. The tetrakis[3,5-bis(trifluoromethyl)phenyl]borate counter-ion lies on a crystallographic twofold axis of symmetry. The trifluoromethyl groups are disordered and were refined with partial occupancies.

Received 9 October 2003 Accepted 4 November 2003 Online 19 December 2003

## Comment

The structure of the title compound, (I) (Fig. 1), consists of (2trimethylsilylethyl)dimethylamine hydrogen-bonded to its ammonium ion, the combined cation lying on a crystallographic inversion centre. The staggered  $N \cdots H \cdots N$ hydrogen bond is characterized by an N-H distance of 1.347 (2) Å and an  $N \cdots N$  distance of 2.695 (3) Å. This compares with the quinicludine-quinuclidinium cation, which has an  $N \cdots N$  separation of 2.689 Å (Mareque Rivas & Brammer, 1999). The counter-ion lies on a twofold axis of symmetry.



## **Experimental**

The title compound, (I), was isolated as a by-product from the reaction of a catalytic quantity of triphenylcarbenium tetrakis[3,5-bis(trifluoromethyl)phenyl]borate, (II), with (2-trimethylsilylethyl)-dimethylamine, (III), in triethylsilane, and presumably arises as a result of the presence of adventitious water in the reaction medium.

#### Crystal data $C_{14}H_{39}N_2Si_2^+ \cdot C_{32}H_{12}F_{24}B^ D_r = 1.401 \text{ Mg m}^{-3}$ $M_r = 1154.88$ Mo Ka radiation Orthorhombic, Pbcn Cell parameters from 6546 $a = 17.1047 (11) \text{ \AA}$ reflections b = 16.8652(11) Å $\theta=2.4{-}24.8^\circ$ $\mu=0.18~\mathrm{mm}^{-1}$ c = 18.9828 (12) Å V = 5476.0 (6) Å<sup>2</sup> T = 130.0 (2) K Z = 4Block, colourless

 $\odot$  2004 International Union of Crystallography Printed in Great Britain – all rights reserved

# organic papers

 $0.30 \times 0.30 \times 0.15 \text{ mm}$ 

## Data collection

4835 independent reflections Bruker SMART CCD area-detector diffractometer 3972 reflections with  $I > 2\sigma(I)$  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  $T_{\rm min}=0.776,\ T_{\rm max}=0.970$ 29606 measured reflections Refinement Refinement on  $F^2$ 

 $R[F^2 > 2\sigma(F^2)] = 0.046$ wR(F<sup>2</sup>) = 0.126 S=1.044835 reflections 407 parameters H-atom parameters constrained

 $R_{\rm int} = 0.038$  $\theta_{\rm max} = 25.0^{\circ}$  $h = -19 \rightarrow 20$  $k = -20 \rightarrow 19$  $l=-12\rightarrow 22$  $w = 1/[\sigma^2(F_o^2) + (0.0645P)^2$ + 2.2182P] where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\rm max} = 0.010$  $\Delta \rho_{\rm max} = 0.30 \text{ e } \text{\AA}^{-3}$ 

 $\Delta \rho_{\rm min} = -0.20 \text{ e} \text{ Å}^{-3}$ 



### Figure 1

Displacement ellipsoid plot of (I), ellipsoids are at the 20% probability level. Hydrogen atoms have been omitted.

The authors acknowledge financial support from The University of Melbourne.

## References

Bruker (1997). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.

Bruker (1999). SAINT. Version 6.02. Bruker AXS Inc., Madison, Wisconsin, USA.

Bruker (2000). SMART. Version 5.55. Bruker AXS Inc., Madison, Wisconsin, USA.

Mareque Rivas, J. C. & Brammer, L. (1999). Coord. Chem. Rev. 183, 43-80. Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

H atoms, other than the one at the center of the cation, were positioned geometrically and refined as riding, with C-H distances of 0.93–0.97 Å and  $U_{iso}(H) = 1.2U_{eq}(C)$ . Each of the four CF<sub>3</sub> groups was refined as disordered over two orientations, with occupancies 0.932:0.068(5), 0.935:0.065 (4), 0.633:0.367 (13) and 0.958:0.042 (4). Isotropic displacement parameters were refined for the F atoms with low site-occupancy. 'Same distance' constraints were applied to the disordered CF<sub>3</sub> groups to keep geometries reasonable.

Data collection: SMART (Bruker, 2000); cell refinement: SMART; data reduction: SAINT (Bruker, 1999); program(s) used to solve structure: SHELXTL (Bruker, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.